



Breaking the 90 % Barrier: CESAR1 performance in achieving Net-Zero Goals

Diego Morlando, Hanna K. Knuutila

Department of Chemical Engineering, NTNU, Trondheim, Norway

Corresponding author's e-mail address: hanna.knuutila@ntnu.no

Topic(s): Post-combustion capture, Absorption.

Keywords: *Post-combustion carbon capture, Amine-based CO₂ capture, CESAR1, High Capture rate.*

ABSTRACT

CO₂ capture plays a fundamental role in achieving the climate change goals set by the Paris Agreement 2015. Capturing 90% of the CO₂ being emitted from industrial sources is not enough, (Brandl et al., 2021). There is a need to design, optimize and qualify technologies that achieve higher capture rates. The use of chemical solvents, particularly amine-based ones, represents the most advanced technology for Post-Combustion Carbon Capture, (Dutcher et al., 2015).

The CESAR1 solvent, an aqueous blend of 2-amino-2-methyl-1-propanol (AMP) and piperazine (PZ), stands out as an open-source solvent among the others for its low energy consumption and slow degradation rate, (Morlando et al., 2024).

In the pilot tests using the CESAR1 solvent, the most common capture rate aimed for is around 90% and few studies aimed to investigate the energy penalty for a higher CO₂ capture rate (98%), (Morlando et al., 2024). Simulation tools can be employed to investigate the effects of different parameters on the amount of CO₂ captured and the associated energy costs.

In this work, we developed a rate-based process model for CO₂ capture using the CESAR1 solvent in Aspen Plus v14. The thermodynamic package and the process model have been validated using experimental data. For example, the CO₂ solubility model predictions have been compared to the data collected by Hartono et al. (2021) and Bröder et al. (2011) as shown in Figure 1. The model accurately predicts the CO₂ solubility over the temperature range from 40 to 120 °C.



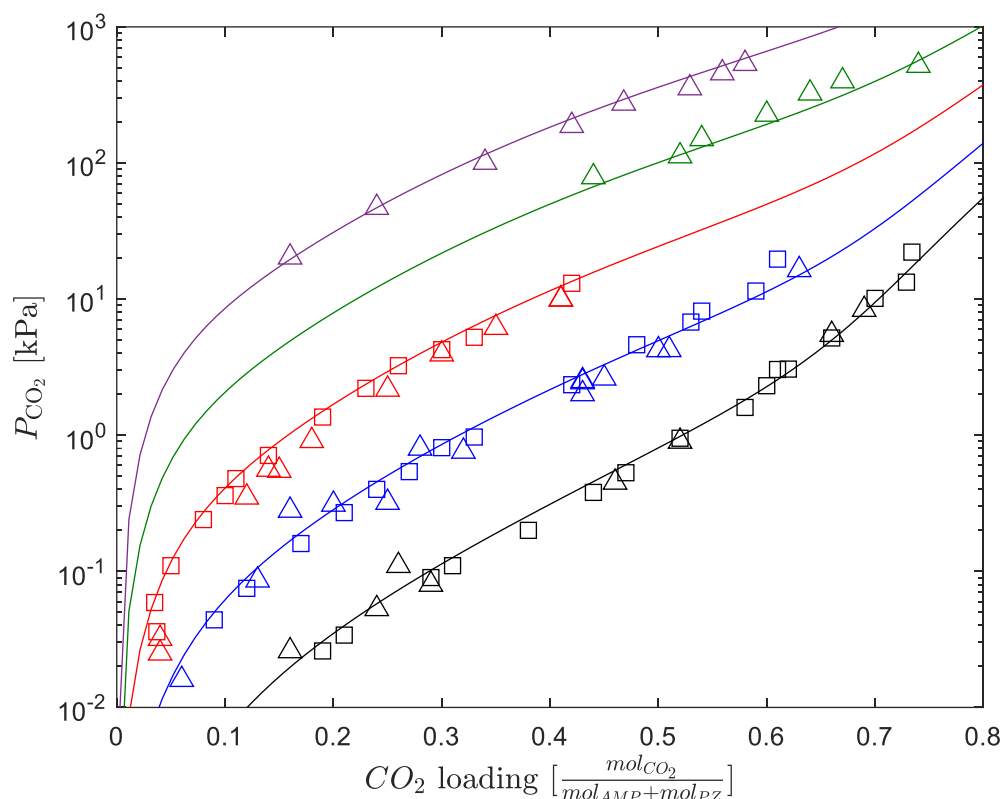


Figure 1: Model prediction of the CO_2 solubility over the CESAR1 solution, (3 M AMP + 1.5 M PZ). ($\square$ Hartono et al. (2021), Δ Bröder et al. (2011)). (Black 40°C, Blue 60 °C, Red 80 °C, Green °100 C, Purple 120°C)

The model will be used to simulate the CO_2 absorption process in a range of CO_2 capture from 90 % to higher than >99% and for different CO_2 concentrations in the flue gas, from 3 vol% to 18 vol%. This study will provide a broad understanding of the CESAR1 solvent performance at high capture rates for different industrial scenarios.

ACKNOWLEDGEMENTS

This research has received funding from the European Union's Horizon Europe research and innovation programme under grant agreement No. 101096521.

REFERENCES

- Brandl, P., Bui, M., Hallett, J. P., & Mac Dowell, N. (2021). Beyond 90% capture: Possible, but at what cost? *International Journal of Greenhouse Gas Control*, 105, 103239. <https://doi.org/https://doi.org/10.1016/j.ijggc.2020.103239>
- Bröder, P., Grimstvedt, A., Mejdell, T., & Svendsen, H. F. (2011). CO_2 capture into aqueous solutions of piperazine activated 2-amino-2-methyl-1-propanol. *Chemical Engineering Science*, 66(23), 6193-6198. <https://doi.org/10.1016/j.ces.2011.08.051>

- Dutcher, B., Fan, M., & Russell, A. G. (2015). Amine-Based CO₂ Capture Technology Development from the Beginning of 2013—A Review. *ACS Applied Materials & Interfaces*, 7(4), 2137-2148. <https://doi.org/10.1021/am507465f>
- Hartono, A., Ahmad, R., Svendsen, H. F., & Knuutila, H. K. (2021). New solubility and heat of absorption data for CO₂ in blends of 2-amino-2-methyl-1-propanol (AMP) and Piperazine (PZ) and a new eNRTL model representation. *Fluid Phase Equilibria*, 550. <https://doi.org/10.1016/j.fluid.2021.113235>
- Morlando, D., Buvik, V., Delic, A., Hartono, A., Svendsen, H. F., Kvamsdal, H. M., da Silva, E. F., & Knuutila, H. K. (2024). Available data and knowledge gaps of the CESAR1 solvent system. *Carbon Capture Science & Technology*, 13, 100290. <https://doi.org/https://doi.org/10.1016/j.ccst.2024.100290>